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## N-[(9-Ethyl-9H-carbazol-3-yl)methylidene]-3,4-dimethylisoxazol-5-amine

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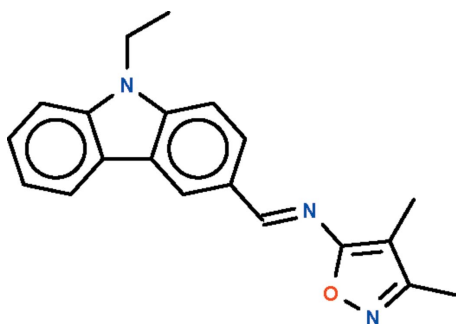
Received 17 June 2010; accepted 12 July 2010

Key indicators: single-crystal X-ray study;  $T = 100$  K; mean  $\sigma(\text{C}-\text{C}) = 0.007$  Å;  $R$  factor = 0.089;  $wR$  factor = 0.262; data-to-parameter ratio = 12.7.

The azomethine double bond in the title Schiff base,  $\text{C}_{20}\text{H}_{19}\text{N}_3\text{O}$ , has an *E* configuration. The 13-membered carbazolyl fused ring system [r.m.s. deviation = 0.023 (9) Å] is nearly coplanar with the five-membered pyrazole ring [r.m.s. deviation = 0.003 (4) Å]; the dihedral angle between the two systems is 10.8 (2)°. The crystal studied was a non-merohedral twin having a 35% minor component.

### Related literature

For the synthesis and spectroscopic characterization of the title compound, see: Asiri *et al.* (2010). For the treatment of non-merohedral twins, see: Spek (2009).



### Experimental

#### Crystal data

$\text{C}_{20}\text{H}_{19}\text{N}_3\text{O}$	$V = 1605.0 (3) \text{ \AA}^3$
$M_r = 317.38$	$Z = 4$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 8.0575 (9) \text{ \AA}$	$\mu = 0.08 \text{ mm}^{-1}$
$b = 13.4483 (15) \text{ \AA}$	$T = 100 \text{ K}$
$c = 14.8488 (16) \text{ \AA}$	$0.20 \times 0.15 \times 0.05 \text{ mm}$
$\beta = 94.049 (2)^\circ$	

#### Data collection

Bruker SMART APEX diffractometer	2811 independent reflections
11993 measured reflections	2286 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.057$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.089$	221 parameters
$wR(F^2) = 0.262$	H-atom parameters constrained
$S = 1.11$	$\Delta\rho_{\text{max}} = 0.55 \text{ e \AA}^{-3}$
2811 reflections	$\Delta\rho_{\text{min}} = -0.54 \text{ e \AA}^{-3}$

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINTE* (Bruker, 2009); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2010).

We thank King Abdul Aziz University and the University of Malaya for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: RK2213).

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**supplementary materials**

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## *N*-[(9-Ethyl-9*H*-carbazol-3-yl)methylidene]-3,4-dimethylisoxazol-5-amine

A. M. Asiri, S. A. Khan, K. W. Tan and S. W. Ng

### Comment

We have recently reported the synthesis and characterization of this Schiff base (Asiri *et al.*, 2010). The azomethine double-bond of the Schiff base (Scheme, Fig. 1) has an *E*-configuration. The 13-membered carbazolyl fused-ring (r.m.s. deviation 0.023 (9)Å) is nearly coplanar with 5-membered pyrazolyl ring (r.m.s. deviation 0.003 (4)Å) (dihedral angle between the two systems 10.8 (2)°).

### Experimental

Crystals of the compound were synthesized as reported (Asiri *et al.*, 2010).

### Refinement

Carbon-bound H-atoms were placed in calculated positions [C—H 0.95 to 0.98Å with  $U_{\text{iso}}(\text{H})$  1.2 to 1.5 $U_{\text{eq}}(\text{C})$ ] and were included in the refinement in the riding model approximation.

The crystal studied is a non-merohedral twin; *PLATON* (Spek, 2009) gave the twin law as (-1 0 0, 0 -1 0, 0.260 0 -1). The de-twinning option lowered the *R* index from 17% to a more acceptable level. However, the twinning affected the refinement so that the thermal ellipsoids, particularly those of the heterocyclic ring, were somewhat large.

### Figures

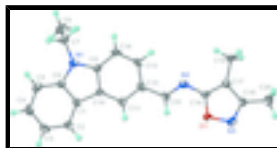


Fig. 1. The molecular structure of title compound with the atom numbering scheme. Displacement ellipsoids are drawn at 70% probability level. Hydrogen atoms are drawn as a small spheres of arbitrary radius.

## *N*-[(9-Ethyl-9*H*-carbazol-3-yl)methylidene]-3,4- dimethylisoxazol-5-amine

### Crystal data

C<sub>20</sub>H<sub>19</sub>N<sub>3</sub>O

$M_r = 317.38$

Monoclinic,  $P2_1/c$

Hall symbol: -P 2ybc

$a = 8.0575$  (9) Å

$b = 13.4483$  (15) Å

$c = 14.8488$  (16) Å

$\beta = 94.049$  (2)°

$F(000) = 672$

$D_x = 1.313$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 3958 reflections

$\theta = 2.5$ – $27.4$ °

$\mu = 0.08$  mm<sup>-1</sup>

$T = 100$  K

Prism, yellow

# supplementary materials

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$V = 1605.0(3) \text{ \AA}^3$   
 $Z = 4$

$0.20 \times 0.15 \times 0.05 \text{ mm}$

## Data collection

Bruker SMART APEX diffractometer	2286 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube	$R_{\text{int}} = 0.057$
graphite	$\theta_{\text{max}} = 25.0^\circ$ , $\theta_{\text{min}} = 2.1^\circ$
$\omega$ scans	$h = -9 \rightarrow 8$
11993 measured reflections	$k = -15 \rightarrow 15$
2811 independent reflections	$l = -17 \rightarrow 17$

## Refinement

Refinement on $F^2$	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.089$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.262$	H-atom parameters constrained
$S = 1.11$	$w = 1/[\sigma^2(F_o^2) + (0.1086P)^2 + 4.6936P]$
2811 reflections	where $P = (F_o^2 + 2F_c^2)/3$
221 parameters	$(\Delta/\sigma)_{\text{max}} = 0.001$
0 restraints	$\Delta\rho_{\text{max}} = 0.55 \text{ e \AA}^{-3}$
	$\Delta\rho_{\text{min}} = -0.54 \text{ e \AA}^{-3}$

## Special details

**Geometry.** All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

## Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.2910 (4)	0.3151 (2)	0.1571 (2)	0.0238 (8)
N1	1.1980 (5)	0.1389 (3)	0.4076 (3)	0.0219 (9)
N2	0.4716 (5)	0.1815 (3)	0.2053 (3)	0.0216 (9)
N3	0.1279 (5)	0.3252 (3)	0.1158 (3)	0.0253 (9)
C1	1.1728 (6)	0.3018 (3)	0.3684 (3)	0.0187 (10)
C2	1.2291 (6)	0.3999 (3)	0.3616 (3)	0.0215 (10)

H2	1.1602	0.4489	0.3320	0.026*
C3	1.3860 (6)	0.4244 (3)	0.3984 (3)	0.0257 (11)
H3	1.4247	0.4909	0.3947	0.031*
C4	1.4879 (6)	0.3522 (4)	0.4411 (3)	0.0257 (11)
H4	1.5947	0.3708	0.4667	0.031*
C5	1.4375 (6)	0.2540 (4)	0.4472 (3)	0.0248 (11)
H5	1.5084	0.2049	0.4753	0.030*
C6	1.2794 (6)	0.2302 (3)	0.4105 (3)	0.0185 (10)
C7	1.2634 (6)	0.0460 (3)	0.4467 (3)	0.0239 (11)
H7A	1.1712	0.0081	0.4710	0.029*
H7B	1.3453	0.0614	0.4976	0.029*
C8	1.3456 (7)	-0.0178 (4)	0.3792 (4)	0.0336 (13)
H8A	1.3818	-0.0805	0.4078	0.050*
H8B	1.4423	0.0174	0.3583	0.050*
H8C	1.2662	-0.0315	0.3277	0.050*
C9	1.0436 (6)	0.1505 (3)	0.3641 (3)	0.0175 (10)
C10	1.0204 (6)	0.2514 (3)	0.3393 (3)	0.0180 (10)
C11	0.8678 (6)	0.2813 (3)	0.2978 (3)	0.0193 (10)
H11	0.8495	0.3491	0.2821	0.023*
C12	0.7420 (6)	0.2112 (3)	0.2796 (3)	0.0212 (10)
C13	0.7718 (6)	0.1102 (3)	0.3024 (3)	0.0212 (10)
H13	0.6869	0.0627	0.2875	0.025*
C14	0.9185 (6)	0.0784 (3)	0.3453 (3)	0.0228 (11)
H14	0.9354	0.0107	0.3616	0.027*
C15	0.5842 (6)	0.2432 (3)	0.2365 (3)	0.0205 (10)
H15	0.5625	0.3124	0.2311	0.025*
C16	0.3229 (6)	0.2162 (3)	0.1664 (3)	0.0186 (10)
C17	0.1911 (6)	0.1619 (3)	0.1324 (3)	0.0207 (10)
C18	0.0728 (6)	0.2352 (3)	0.1022 (3)	0.0197 (10)
C19	0.1746 (6)	0.0520 (3)	0.1276 (3)	0.0253 (11)
H19A	0.2802	0.0211	0.1493	0.038*
H19B	0.1463	0.0320	0.0650	0.038*
H19C	0.0866	0.0304	0.1655	0.038*
C20	-0.0982 (6)	0.2179 (4)	0.0589 (3)	0.0280 (11)
H20A	-0.1665	0.1847	0.1020	0.042*
H20B	-0.0911	0.1758	0.0054	0.042*
H20C	-0.1488	0.2818	0.0410	0.042*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0252 (18)	0.0164 (16)	0.0296 (18)	-0.0002 (14)	0.0004 (14)	0.0000 (13)
N1	0.026 (2)	0.0186 (19)	0.021 (2)	0.0010 (17)	0.0022 (17)	0.0028 (15)
N2	0.023 (2)	0.0176 (19)	0.024 (2)	-0.0013 (17)	0.0046 (17)	0.0003 (15)
N3	0.024 (2)	0.024 (2)	0.028 (2)	0.0035 (18)	0.0010 (18)	0.0026 (17)
C1	0.022 (2)	0.019 (2)	0.015 (2)	-0.0006 (19)	0.0041 (18)	-0.0030 (17)
C2	0.021 (2)	0.016 (2)	0.027 (2)	0.0013 (19)	0.002 (2)	-0.0006 (18)
C3	0.024 (3)	0.019 (2)	0.034 (3)	-0.003 (2)	0.004 (2)	-0.003 (2)

## supplementary materials

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C4	0.021 (2)	0.031 (3)	0.024 (2)	-0.001 (2)	0.001 (2)	-0.004 (2)
C5	0.029 (3)	0.027 (2)	0.019 (2)	0.009 (2)	0.001 (2)	0.0007 (19)
C6	0.021 (2)	0.018 (2)	0.017 (2)	0.0017 (19)	0.0020 (18)	-0.0010 (17)
C7	0.031 (3)	0.016 (2)	0.025 (2)	0.003 (2)	-0.002 (2)	0.0036 (18)
C8	0.037 (3)	0.027 (3)	0.037 (3)	0.009 (2)	0.000 (2)	0.002 (2)
C9	0.021 (2)	0.016 (2)	0.015 (2)	0.0030 (18)	0.0046 (19)	0.0001 (17)
C10	0.024 (3)	0.014 (2)	0.016 (2)	-0.0009 (18)	0.0040 (19)	-0.0010 (16)
C11	0.023 (3)	0.017 (2)	0.018 (2)	0.0014 (19)	0.0032 (19)	-0.0007 (17)
C12	0.024 (3)	0.021 (2)	0.019 (2)	-0.003 (2)	0.0043 (19)	-0.0026 (18)
C13	0.025 (3)	0.017 (2)	0.022 (2)	-0.0061 (19)	0.005 (2)	-0.0020 (18)
C14	0.030 (3)	0.013 (2)	0.025 (2)	-0.001 (2)	0.005 (2)	0.0007 (18)
C15	0.021 (2)	0.017 (2)	0.024 (2)	-0.0023 (19)	0.005 (2)	-0.0019 (18)
C16	0.023 (2)	0.014 (2)	0.019 (2)	0.0001 (18)	0.0054 (19)	0.0019 (17)
C17	0.021 (2)	0.019 (2)	0.022 (2)	-0.0016 (19)	0.0039 (19)	-0.0006 (18)
C18	0.021 (2)	0.020 (2)	0.018 (2)	-0.0012 (19)	0.0040 (19)	0.0003 (18)
C19	0.024 (3)	0.016 (2)	0.036 (3)	-0.002 (2)	-0.003 (2)	0.0011 (19)
C20	0.025 (3)	0.033 (3)	0.026 (3)	0.000 (2)	0.000 (2)	0.001 (2)

### Geometric parameters (Å, °)

O1—C16	1.360 (5)	C8—H8B	0.9800
O1—N3	1.417 (5)	C8—H8C	0.9800
N1—C9	1.369 (6)	C9—C14	1.412 (6)
N1—C6	1.391 (6)	C9—C10	1.416 (6)
N1—C7	1.461 (6)	C10—C11	1.395 (6)
N2—C15	1.291 (6)	C11—C12	1.397 (6)
N2—C16	1.375 (6)	C11—H11	0.9500
N3—C18	1.300 (6)	C12—C13	1.416 (6)
C1—C2	1.400 (6)	C12—C15	1.448 (7)
C1—C6	1.407 (6)	C13—C14	1.371 (7)
C1—C10	1.443 (6)	C13—H13	0.9500
C2—C3	1.381 (7)	C14—H14	0.9500
C2—H2	0.9500	C15—H15	0.9500
C3—C4	1.395 (7)	C16—C17	1.357 (6)
C3—H3	0.9500	C17—C18	1.422 (6)
C4—C5	1.386 (7)	C17—C19	1.485 (6)
C4—H4	0.9500	C18—C20	1.496 (7)
C5—C6	1.387 (7)	C19—H19A	0.9800
C5—H5	0.9500	C19—H19B	0.9800
C7—C8	1.507 (7)	C19—H19C	0.9800
C7—H7A	0.9900	C20—H20A	0.9800
C7—H7B	0.9900	C20—H20B	0.9800
C8—H8A	0.9800	C20—H20C	0.9800
C16—O1—N3	107.5 (3)	C11—C10—C9	119.2 (4)
C9—N1—C6	109.0 (4)	C11—C10—C1	134.6 (4)
C9—N1—C7	125.1 (4)	C9—C10—C1	106.2 (4)
C6—N1—C7	125.9 (4)	C10—C11—C12	119.7 (4)
C15—N2—C16	120.1 (4)	C10—C11—H11	120.1
C18—N3—O1	105.9 (4)	C12—C11—H11	120.1

C2—C1—C6	119.1 (4)	C11—C12—C13	119.5 (4)
C2—C1—C10	134.0 (4)	C11—C12—C15	119.2 (4)
C6—C1—C10	107.0 (4)	C13—C12—C15	121.3 (4)
C3—C2—C1	119.2 (4)	C14—C13—C12	122.4 (4)
C3—C2—H2	120.4	C14—C13—H13	118.8
C1—C2—H2	120.4	C12—C13—H13	118.8
C2—C3—C4	120.4 (4)	C13—C14—C9	117.3 (4)
C2—C3—H3	119.8	C13—C14—H14	121.4
C4—C3—H3	119.8	C9—C14—H14	121.4
C5—C4—C3	121.8 (5)	N2—C15—C12	122.7 (4)
C5—C4—H4	119.1	N2—C15—H15	118.6
C3—C4—H4	119.1	C12—C15—H15	118.6
C4—C5—C6	117.3 (4)	C17—C16—O1	110.6 (4)
C4—C5—H5	121.3	C17—C16—N2	127.6 (4)
C6—C5—H5	121.3	O1—C16—N2	121.9 (4)
C5—C6—N1	129.3 (4)	C16—C17—C18	103.5 (4)
C5—C6—C1	122.1 (4)	C16—C17—C19	128.2 (4)
N1—C6—C1	108.6 (4)	C18—C17—C19	128.3 (4)
N1—C7—C8	112.7 (4)	N3—C18—C17	112.5 (4)
N1—C7—H7A	109.1	N3—C18—C20	120.4 (4)
C8—C7—H7A	109.1	C17—C18—C20	127.2 (4)
N1—C7—H7B	109.1	C17—C19—H19A	109.5
C8—C7—H7B	109.1	C17—C19—H19B	109.5
H7A—C7—H7B	107.8	H19A—C19—H19B	109.5
C7—C8—H8A	109.5	C17—C19—H19C	109.5
C7—C8—H8B	109.5	H19A—C19—H19C	109.5
H8A—C8—H8B	109.5	H19B—C19—H19C	109.5
C7—C8—H8C	109.5	C18—C20—H20A	109.5
H8A—C8—H8C	109.5	C18—C20—H20B	109.5
H8B—C8—H8C	109.5	H20A—C20—H20B	109.5
N1—C9—C14	128.9 (4)	C18—C20—H20C	109.5
N1—C9—C10	109.3 (4)	H20A—C20—H20C	109.5
C14—C9—C10	121.8 (4)	H20B—C20—H20C	109.5
C16—O1—N3—C18	-0.6 (5)	C2—C1—C10—C9	178.0 (5)
C6—C1—C2—C3	-1.7 (6)	C6—C1—C10—C9	-1.0 (5)
C10—C1—C2—C3	179.5 (5)	C9—C10—C11—C12	-1.5 (6)
C1—C2—C3—C4	0.6 (7)	C1—C10—C11—C12	-179.1 (5)
C2—C3—C4—C5	0.9 (7)	C10—C11—C12—C13	-0.8 (6)
C3—C4—C5—C6	-1.2 (7)	C10—C11—C12—C15	180.0 (4)
C4—C5—C6—N1	-179.6 (4)	C11—C12—C13—C14	2.5 (7)
C4—C5—C6—C1	0.1 (7)	C15—C12—C13—C14	-178.3 (4)
C9—N1—C6—C5	-179.6 (4)	C12—C13—C14—C9	-1.8 (7)
C7—N1—C6—C5	2.0 (7)	N1—C9—C14—C13	178.3 (4)
C9—N1—C6—C1	0.6 (5)	C10—C9—C14—C13	-0.6 (6)
C7—N1—C6—C1	-177.8 (4)	C16—N2—C15—C12	178.9 (4)
C2—C1—C6—C5	1.3 (6)	C11—C12—C15—N2	169.3 (4)
C10—C1—C6—C5	-179.6 (4)	C13—C12—C15—N2	-9.9 (7)
C2—C1—C6—N1	-178.9 (4)	N3—O1—C16—C17	0.9 (5)
C10—C1—C6—N1	0.2 (5)	N3—O1—C16—N2	-179.4 (4)

## supplementary materials

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C9—N1—C7—C8	86.7 (6)	C15—N2—C16—C17	-178.4 (4)
C6—N1—C7—C8	-95.2 (5)	C15—N2—C16—O1	2.0 (6)
C6—N1—C9—C14	179.8 (4)	O1—C16—C17—C18	-0.9 (5)
C7—N1—C9—C14	-1.9 (7)	N2—C16—C17—C18	179.5 (4)
C6—N1—C9—C10	-1.2 (5)	O1—C16—C17—C19	178.9 (4)
C7—N1—C9—C10	177.1 (4)	N2—C16—C17—C19	-0.7 (8)
N1—C9—C10—C11	-176.9 (4)	O1—N3—C18—C17	0.1 (5)
C14—C9—C10—C11	2.2 (6)	O1—N3—C18—C20	-179.9 (4)
N1—C9—C10—C1	1.4 (5)	C16—C17—C18—N3	0.5 (5)
C14—C9—C10—C1	-179.6 (4)	C19—C17—C18—N3	-179.3 (5)
C2—C1—C10—C11	-4.2 (9)	C16—C17—C18—C20	-179.5 (4)
C6—C1—C10—C11	176.8 (5)	C19—C17—C18—C20	0.7 (8)



Fig. 1

