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## Structure Reports

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# Chlorido(2-methyl-4-oxo-4H-pyran-3-olato- $\kappa^2O^3,O^4$ )(1,10-phenanthroline- $\kappa^2N,N'$ )copper(II)

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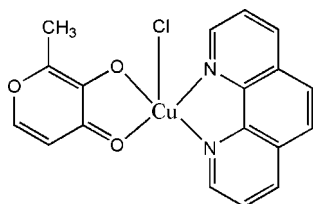
Received 25 July 2008; accepted 28 July 2008

Key indicators: single-crystal X-ray study;  $T = 100$  K; mean  $\sigma(C-C) = 0.006$  Å;  $R$  factor = 0.056;  $wR$  factor = 0.169; data-to-parameter ratio = 15.1.

The copper(II) atoms in the two independent molecules of the title compound,  $[Cu(C_6H_5O_3)Cl(C_{12}H_8N_2)]$ , both adopt square-pyramidal geometries. The two coordinating atoms of the two heterocyclic ligands comprise the square plane, and the chlorine atom occupies the apical position of the coordination environment.

## Related literature

For the structure of aquachlorido(maltolato)copper hydrate, which adopts a chlorido-bridged chain structure, see: Odoko *et al.* (2002).



## Experimental

### Crystal data

$[Cu(C_6H_5O_3)Cl(C_{12}H_8N_2)]$   
 $M_r = 404.29$   
 Triclinic,  $P\bar{1}$   
 $a = 9.0043$  (2) Å  
 $b = 12.0599$  (2) Å  
 $c = 14.4655$  (3) Å  
 $\alpha = 77.412$  (1)°  
 $\beta = 87.380$  (1)°

$\gamma = 84.565$  (1)°  
 $V = 1525.66$  (5) Å<sup>3</sup>  
 $Z = 4$   
 Mo  $K\alpha$  radiation  
 $\mu = 1.63$  mm<sup>-1</sup>  
 $T = 100$  (2) K  
 $0.15 \times 0.05 \times 0.05$  mm

### Data collection

Bruker SMART APEX  
 diffractometer  
 Absorption correction: multi-scan  
 (SADABS; Sheldrick, 1996)  
 $T_{min} = 0.792$ ,  $T_{max} = 0.923$

14221 measured reflections  
 6825 independent reflections  
 5053 reflections with  $I > 2\sigma(I)$   
 $R_{int} = 0.041$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.056$   
 $wR(F^2) = 0.169$   
 $S = 1.04$   
 6825 reflections

453 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{max} = 2.04$  e Å<sup>-3</sup>  
 $\Delta\rho_{min} = -1.26$  e Å<sup>-3</sup>

**Table 1**

Selected bond lengths (Å).

Cu1—N1	2.001 (3)	Cu2—O4	1.989 (3)
Cu1—N2	2.010 (3)	Cu2—O5	1.924 (3)
Cu1—O1	1.979 (3)	Cu2—N3	1.999 (3)
Cu1—O2	1.920 (3)	Cu2—N4	2.014 (3)
Cu1—Cl1	2.540 (1)	Cu2—Cl2	2.524 (1)

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINTE* (Bruker, 2007); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2008).

We thank MOSTI (eSc Project No. 02-02-11-SF0033) for supporting this study; KWT thanks the Ministry of Higher Education for a SLAI scholarship.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU2443).

## References

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**supplementary materials**

*Acta Cryst.* (2008). E64, m1104 [ doi:10.1107/S1600536808023829 ]

## Chlorido(2-methyl-4-oxo-4*H*-pyran-3-olato- $\kappa^2 O^3, O^4$ )(1,10-phenanthroline- $\kappa^2 N, N'$ )copper(II)

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### Comment

There are few crystal structure studies of metal derivatives of maltol. A monochlorocopper(II) derivative is known; the deprotonated ligand chelates to the metal atom; adjacent molecules are linked into a chain by chloride bridges (Odoko *et al.*, 2002). In the present 1,10-phenanthroline adduct (Scheme I), the deprotonated ligand and the *N*-heterocycle both chelate to the copper atom, which shows square pyramidal coordination (Fig. 1 and Table 1). There are two independent molecules in the asymmetric unit.

### Experimental

Solid maltol (0.20 g, 1.6 mmol) was added to a sodium hydroxide solution (0.60 g, 1.5 mmol, 20 ml) followed by 1,10-phenanthroline (0.27 g, 1.5 mmol) dissolved in methanol (5 ml). Copper(II) chloride dihydrate ((0.26 g, 1.5 mmol) dissolved in water (10 ml) was added. The green solution was set aside for the growth of crystals.

### Refinement

Carbon-bound H-atoms were placed in calculated positions (C—H 0.95 to 0.98 Å) and were included in the refinement in the riding model approximation, with  $U_{\text{iso}}(\text{H})$  set to 1.2  $U_{\text{eq}}(\text{C})$ . The final difference Fourier map had a large peak/deep hole at about 1 Å from Cu1.

### Figures

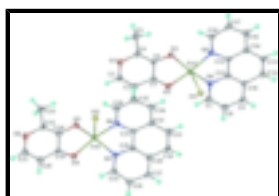


Fig. 1. Thermal ellipsoid (Barbour, 2001) plot of the two independent molecules of  $\text{CuCl}(\text{C}_{12}\text{H}_8\text{N}_2)(\text{C}_6\text{H}_5\text{O}_3)$  at the 70% probability level. Hydrogen atoms are drawn as spheres of arbitrary radii.

### (I)

#### Crystal data

$[\text{Cu}(\text{C}_6\text{H}_5\text{O}_3)\text{Cl}(\text{C}_{12}\text{H}_8\text{N}_2)]$

$M_r = 404.29$

Triclinic, *PT*

Hall symbol: -P 1

$a = 9.0043$  (2) Å

$Z = 4$

$F_{000} = 820$

$D_x = 1.760$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation

$\lambda = 0.71073$  Å

Cell parameters from 4344 reflections

## supplementary materials

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$b = 12.0599 (2) \text{ \AA}$   
 $c = 14.4655 (3) \text{ \AA}$   
 $\alpha = 77.412 (1)^\circ$   
 $\beta = 87.380 (1)^\circ$   
 $\gamma = 84.565 (1)^\circ$   
 $V = 1525.66 (5) \text{ \AA}^3$

$\theta = 2.3\text{--}28.2^\circ$   
 $\mu = 1.63 \text{ mm}^{-1}$   
 $T = 100 (2) \text{ K}$   
Irregular block, green  
 $0.15 \times 0.05 \times 0.05 \text{ mm}$

### Data collection

Bruker SMART APEX diffractometer  
Radiation source: fine-focus sealed tube  
Monochromator: graphite  
 $T = 100(2) \text{ K}$   
 $\omega$  scans  
Absorption correction: Multi-scan (SADABS; Sheldrick, 1996)  
 $T_{\min} = 0.792$ ,  $T_{\max} = 0.923$   
14221 measured reflections

6825 independent reflections  
5053 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.041$   
 $\theta_{\text{max}} = 27.5^\circ$   
 $\theta_{\text{min}} = 1.4^\circ$   
 $h = -11 \rightarrow 11$   
 $k = -15 \rightarrow 15$   
 $l = -18 \rightarrow 18$

### Refinement

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.056$   
 $wR(F^2) = 0.169$   
 $S = 1.04$   
6825 reflections  
453 parameters  
Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map  
Hydrogen site location: inferred from neighbouring sites  
H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.1077P)^2 + 0.3956P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} = 0.001$   
 $\Delta\rho_{\text{max}} = 2.04 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -1.26 \text{ e \AA}^{-3}$   
Extinction correction: none

### Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Cu1	0.85441 (5)	0.77700 (4)	0.28183 (3)	0.01441 (15)
Cu2	0.57072 (5)	0.17397 (4)	0.21656 (3)	0.01458 (15)
O1	0.7016 (3)	0.6822 (2)	0.25382 (19)	0.0172 (6)
O2	0.8556 (3)	0.6797 (2)	0.40608 (18)	0.0165 (6)
O3	0.6602 (3)	0.4274 (2)	0.48535 (19)	0.0169 (6)
O4	0.4569 (3)	0.0598 (2)	0.17636 (19)	0.0169 (6)
O5	0.6122 (3)	0.0568 (2)	0.32809 (18)	0.0171 (6)
O6	0.4537 (3)	-0.2134 (2)	0.39503 (19)	0.0180 (6)
N1	0.8155 (3)	0.9036 (3)	0.1685 (2)	0.0150 (7)
N2	0.9678 (3)	0.8943 (3)	0.3209 (2)	0.0152 (7)
N3	0.5841 (4)	0.2733 (3)	0.0869 (2)	0.0160 (7)

N4	0.7330 (3)	0.2675 (3)	0.2404 (2)	0.0140 (6)
C1	0.6830 (4)	0.6016 (3)	0.3256 (3)	0.0142 (7)
C2	0.5859 (4)	0.5143 (3)	0.3298 (3)	0.0168 (8)
H2A	0.5254	0.5147	0.2776	0.020*
C3	0.5805 (4)	0.4310 (3)	0.4083 (3)	0.0179 (8)
H3A	0.5173	0.3720	0.4094	0.021*
C4	0.7526 (4)	0.5118 (3)	0.4863 (3)	0.0161 (8)
C5	0.8287 (5)	0.4966 (4)	0.5774 (3)	0.0216 (9)
H5A	0.7598	0.5234	0.6239	0.032*
H5B	0.9162	0.5406	0.5679	0.032*
H5C	0.8603	0.4156	0.6007	0.032*
C6	0.7679 (4)	0.5981 (3)	0.4088 (3)	0.0147 (8)
C7	0.7350 (4)	0.9050 (4)	0.0933 (3)	0.0185 (8)
H7A	0.6753	0.8438	0.0945	0.022*
C8	0.7352 (5)	0.9933 (4)	0.0127 (3)	0.0198 (8)
H8A	0.6759	0.9919	-0.0397	0.024*
C9	0.8216 (4)	1.0823 (4)	0.0096 (3)	0.0181 (8)
H9A	0.8234	1.1424	-0.0451	0.022*
C10	0.9075 (4)	1.0833 (3)	0.0883 (3)	0.0158 (8)
C11	0.8986 (4)	0.9918 (3)	0.1668 (3)	0.0145 (8)
C12	1.0033 (4)	1.1708 (3)	0.0933 (3)	0.0178 (8)
H12A	1.0108	1.2333	0.0409	0.021*
C13	1.0835 (4)	1.1652 (3)	0.1722 (3)	0.0179 (8)
H13A	1.1480	1.2228	0.1732	0.021*
C14	1.0723 (4)	1.0742 (3)	0.2534 (3)	0.0165 (8)
C15	0.9835 (4)	0.9861 (3)	0.2501 (3)	0.0141 (8)
C16	1.1477 (4)	1.0643 (4)	0.3396 (3)	0.0182 (8)
H16A	1.2096	1.1212	0.3471	0.022*
C17	1.1295 (4)	0.9718 (4)	0.4112 (3)	0.0201 (9)
H17A	1.1781	0.9647	0.4694	0.024*
C18	1.0403 (4)	0.8874 (3)	0.4002 (3)	0.0167 (8)
H18A	1.0312	0.8231	0.4508	0.020*
C19	0.4546 (4)	-0.0287 (3)	0.2423 (3)	0.0143 (8)
C20	0.3761 (4)	-0.1249 (4)	0.2403 (3)	0.0185 (8)
H20A	0.3229	-0.1279	0.1858	0.022*
C21	0.3780 (4)	-0.2119 (4)	0.3169 (3)	0.0191 (8)
H21A	0.3233	-0.2750	0.3154	0.023*
C22	0.5334 (4)	-0.1241 (3)	0.4005 (3)	0.0170 (8)
C23	0.6088 (5)	-0.1400 (4)	0.4921 (3)	0.0203 (9)
H23A	0.6762	-0.2100	0.5018	0.030*
H23B	0.5338	-0.1457	0.5438	0.030*
H23C	0.6662	-0.0747	0.4913	0.030*
C24	0.5369 (4)	-0.0313 (3)	0.3263 (3)	0.0158 (8)
C25	0.5112 (4)	0.2693 (4)	0.0096 (3)	0.0187 (8)
H25A	0.4442	0.2121	0.0132	0.022*
C26	0.5303 (5)	0.3467 (4)	-0.0766 (3)	0.0191 (8)
H26A	0.4799	0.3399	-0.1311	0.023*
C27	0.6228 (4)	0.4331 (3)	-0.0820 (3)	0.0178 (8)
H27A	0.6357	0.4868	-0.1399	0.021*

## supplementary materials

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C28	0.6980 (4)	0.4406 (3)	-0.0005 (3)	0.0168 (8)
C29	0.6754 (4)	0.3571 (3)	0.0821 (3)	0.0148 (8)
C30	0.7935 (4)	0.5271 (3)	0.0033 (3)	0.0183 (8)
H30A	0.8064	0.5861	-0.0512	0.022*
C31	0.8670 (4)	0.5269 (3)	0.0841 (3)	0.0191 (8)
H31A	0.9272	0.5870	0.0854	0.023*
C32	0.8549 (4)	0.4373 (3)	0.1670 (3)	0.0163 (8)
C33	0.7556 (4)	0.3548 (3)	0.1667 (3)	0.0146 (8)
C34	0.9387 (4)	0.4249 (3)	0.2496 (3)	0.0180 (8)
H34A	1.0080	0.4782	0.2540	0.022*
C35	0.9176 (4)	0.3342 (4)	0.3233 (3)	0.0198 (8)
H35A	0.9745	0.3235	0.3787	0.024*
C36	0.8131 (4)	0.2570 (3)	0.3177 (3)	0.0180 (8)
H36A	0.7990	0.1957	0.3702	0.022*
Cl1	1.08237 (10)	0.70256 (8)	0.19629 (6)	0.0169 (2)
Cl2	0.36235 (10)	0.30584 (8)	0.26606 (6)	0.0171 (2)

### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Cu1	0.0201 (3)	0.0137 (3)	0.0081 (2)	-0.00360 (19)	-0.00410 (17)	0.00211 (18)
Cu2	0.0204 (3)	0.0143 (3)	0.0080 (2)	-0.00389 (19)	-0.00415 (17)	0.00174 (18)
O1	0.0206 (14)	0.0208 (15)	0.0089 (13)	-0.0035 (12)	-0.0059 (10)	0.0014 (11)
O2	0.0207 (14)	0.0191 (14)	0.0090 (13)	-0.0052 (11)	-0.0046 (10)	0.0005 (11)
O3	0.0220 (14)	0.0152 (14)	0.0125 (13)	-0.0042 (11)	-0.0042 (10)	0.0009 (11)
O4	0.0221 (14)	0.0176 (14)	0.0109 (13)	-0.0030 (11)	-0.0055 (10)	-0.0012 (11)
O5	0.0225 (14)	0.0175 (14)	0.0109 (13)	-0.0051 (11)	-0.0066 (10)	0.0004 (11)
O6	0.0242 (14)	0.0161 (14)	0.0127 (14)	-0.0039 (11)	-0.0060 (11)	0.0010 (11)
N1	0.0181 (16)	0.0155 (17)	0.0106 (15)	-0.0019 (13)	-0.0026 (12)	-0.0003 (13)
N2	0.0170 (16)	0.0169 (17)	0.0118 (16)	-0.0026 (13)	-0.0029 (12)	-0.0022 (13)
N3	0.0233 (17)	0.0152 (17)	0.0096 (15)	-0.0033 (13)	-0.0031 (12)	-0.0017 (13)
N4	0.0170 (16)	0.0146 (16)	0.0098 (15)	-0.0013 (13)	-0.0026 (12)	-0.0011 (13)
C1	0.0173 (18)	0.0158 (19)	0.0100 (18)	-0.0015 (15)	-0.0012 (14)	-0.0034 (15)
C2	0.0183 (19)	0.020 (2)	0.0125 (18)	-0.0014 (16)	-0.0040 (14)	-0.0023 (15)
C3	0.0199 (19)	0.018 (2)	0.0158 (19)	-0.0053 (16)	-0.0002 (15)	-0.0023 (16)
C4	0.0178 (19)	0.017 (2)	0.0120 (18)	-0.0015 (15)	-0.0031 (14)	-0.0006 (15)
C5	0.028 (2)	0.023 (2)	0.0125 (19)	-0.0068 (18)	-0.0092 (15)	0.0022 (16)
C6	0.0153 (18)	0.018 (2)	0.0099 (17)	0.0010 (15)	0.0005 (13)	-0.0034 (15)
C7	0.023 (2)	0.019 (2)	0.0118 (18)	-0.0029 (16)	-0.0053 (14)	0.0015 (15)
C8	0.025 (2)	0.020 (2)	0.0138 (19)	-0.0004 (17)	-0.0062 (15)	-0.0008 (16)
C9	0.023 (2)	0.020 (2)	0.0081 (18)	0.0024 (16)	-0.0032 (14)	0.0030 (15)
C10	0.0185 (19)	0.017 (2)	0.0111 (18)	0.0010 (15)	0.0014 (14)	-0.0032 (15)
C11	0.0178 (18)	0.0148 (19)	0.0103 (18)	0.0004 (15)	-0.0009 (14)	-0.0018 (15)
C12	0.022 (2)	0.0142 (19)	0.0152 (19)	-0.0010 (16)	-0.0018 (15)	0.0012 (15)
C13	0.0193 (19)	0.018 (2)	0.016 (2)	-0.0034 (16)	0.0007 (15)	-0.0026 (16)
C14	0.0193 (19)	0.016 (2)	0.0136 (19)	-0.0006 (15)	-0.0007 (14)	-0.0030 (15)
C15	0.0150 (17)	0.019 (2)	0.0087 (17)	-0.0016 (15)	0.0002 (13)	-0.0035 (15)
C16	0.0195 (19)	0.019 (2)	0.016 (2)	-0.0042 (16)	-0.0035 (15)	-0.0036 (16)

C17	0.024 (2)	0.027 (2)	0.0079 (18)	-0.0028 (17)	-0.0042 (15)	-0.0003 (16)
C18	0.0222 (19)	0.017 (2)	0.0083 (17)	-0.0006 (16)	-0.0052 (14)	0.0026 (15)
C19	0.0181 (18)	0.018 (2)	0.0070 (17)	-0.0006 (15)	-0.0020 (13)	-0.0022 (14)
C20	0.021 (2)	0.021 (2)	0.0139 (19)	-0.0013 (16)	-0.0058 (15)	-0.0037 (16)
C21	0.022 (2)	0.019 (2)	0.017 (2)	-0.0019 (16)	-0.0029 (15)	-0.0027 (16)
C22	0.0211 (19)	0.017 (2)	0.0119 (18)	-0.0032 (16)	0.0014 (14)	-0.0018 (15)
C23	0.029 (2)	0.020 (2)	0.0092 (18)	-0.0022 (17)	-0.0067 (15)	0.0029 (16)
C24	0.0172 (18)	0.0162 (19)	0.0132 (19)	0.0010 (15)	-0.0042 (14)	-0.0019 (15)
C25	0.023 (2)	0.020 (2)	0.0131 (19)	-0.0029 (16)	-0.0060 (15)	-0.0018 (16)
C26	0.024 (2)	0.023 (2)	0.0097 (18)	0.0001 (17)	-0.0049 (14)	-0.0024 (16)
C27	0.025 (2)	0.019 (2)	0.0068 (17)	0.0011 (16)	-0.0041 (14)	0.0031 (15)
C28	0.0168 (18)	0.019 (2)	0.0133 (19)	0.0022 (16)	-0.0023 (14)	-0.0015 (16)
C29	0.0167 (18)	0.0154 (19)	0.0118 (18)	0.0003 (15)	-0.0008 (14)	-0.0027 (15)
C30	0.0199 (19)	0.018 (2)	0.0141 (19)	-0.0019 (16)	0.0020 (15)	0.0036 (15)
C31	0.021 (2)	0.016 (2)	0.019 (2)	-0.0028 (16)	0.0012 (15)	-0.0013 (16)
C32	0.0207 (19)	0.0144 (19)	0.0124 (19)	-0.0006 (15)	0.0009 (14)	-0.0005 (15)
C33	0.0189 (19)	0.0146 (19)	0.0106 (18)	0.0000 (15)	-0.0048 (14)	-0.0033 (14)
C34	0.0209 (19)	0.017 (2)	0.016 (2)	-0.0015 (16)	-0.0053 (15)	-0.0031 (16)
C35	0.021 (2)	0.023 (2)	0.015 (2)	-0.0006 (17)	-0.0075 (15)	-0.0029 (16)
C36	0.025 (2)	0.015 (2)	0.0123 (18)	-0.0006 (16)	-0.0043 (15)	0.0012 (15)
Cl1	0.0197 (5)	0.0172 (5)	0.0129 (4)	-0.0019 (4)	-0.0013 (3)	-0.0008 (4)
Cl2	0.0205 (5)	0.0171 (5)	0.0127 (4)	-0.0019 (4)	-0.0027 (3)	-0.0008 (4)

*Geometric parameters (Å, °)*

Cu1—N1	2.001 (3)	C10—C12	1.440 (5)
Cu1—N2	2.010 (3)	C11—C15	1.441 (5)
Cu1—O1	1.979 (3)	C12—C13	1.364 (5)
Cu1—O2	1.920 (3)	C12—H12A	0.9500
Cu1—Cl1	2.540 (1)	C13—C14	1.429 (5)
Cu2—O4	1.989 (3)	C13—H13A	0.9500
Cu2—O5	1.924 (3)	C14—C15	1.398 (5)
Cu2—N3	1.999 (3)	C14—C16	1.424 (5)
Cu2—N4	2.014 (3)	C16—C17	1.364 (5)
Cu2—Cl2	2.524 (1)	C16—H16A	0.9500
O1—C1	1.275 (5)	C17—C18	1.394 (6)
O2—C6	1.312 (5)	C17—H17A	0.9500
O3—C3	1.343 (5)	C18—H18A	0.9500
O3—C4	1.377 (4)	C19—C20	1.420 (5)
O4—C19	1.268 (5)	C19—C24	1.446 (5)
O5—C24	1.320 (5)	C20—C21	1.349 (6)
O6—C21	1.343 (5)	C20—H20A	0.9500
O6—C22	1.367 (5)	C21—H21A	0.9500
N1—C7	1.331 (5)	C22—C24	1.373 (5)
N1—C11	1.353 (5)	C22—C23	1.483 (5)
N2—C18	1.328 (5)	C23—H23A	0.9800
N2—C15	1.349 (5)	C23—H23B	0.9800
N3—C25	1.334 (5)	C23—H23C	0.9800
N3—C29	1.351 (5)	C25—C26	1.401 (5)

## supplementary materials

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N4—C36	1.335 (5)	C25—H25A	0.9500
N4—C33	1.349 (5)	C26—C27	1.381 (6)
C1—C2	1.420 (5)	C26—H26A	0.9500
C1—C6	1.446 (5)	C27—C28	1.410 (5)
C2—C3	1.344 (5)	C27—H27A	0.9500
C2—H2A	0.9500	C28—C29	1.405 (5)
C3—H3A	0.9500	C28—C30	1.425 (5)
C4—C6	1.366 (5)	C29—C33	1.442 (5)
C4—C5	1.481 (5)	C30—C31	1.368 (6)
C5—H5A	0.9800	C30—H30A	0.9500
C5—H5B	0.9800	C31—C32	1.436 (5)
C5—H5C	0.9800	C31—H31A	0.9500
C7—C8	1.397 (5)	C32—C33	1.400 (5)
C7—H7A	0.9500	C32—C34	1.416 (5)
C8—C9	1.377 (6)	C34—C35	1.372 (6)
C8—H8A	0.9500	C34—H34A	0.9500
C9—C10	1.409 (5)	C35—C36	1.401 (5)
C9—H9A	0.9500	C35—H35A	0.9500
C10—C11	1.406 (5)	C36—H36A	0.9500
O2—Cu1—O1	85.57 (11)	C10—C12—H12A	119.6
O2—Cu1—N1	165.20 (13)	C12—C13—C14	121.3 (4)
O1—Cu1—N1	95.47 (12)	C12—C13—H13A	119.4
O2—Cu1—N2	93.73 (12)	C14—C13—H13A	119.4
O1—Cu1—N2	166.51 (13)	C15—C14—C16	115.9 (3)
N1—Cu1—N2	81.81 (13)	C15—C14—C13	119.3 (4)
O2—Cu1—C11	104.77 (9)	C16—C14—C13	124.7 (4)
O1—Cu1—C11	101.17 (9)	N2—C15—C14	124.8 (3)
N1—Cu1—C11	89.55 (10)	N2—C15—C11	115.8 (3)
N2—Cu1—C11	92.05 (10)	C14—C15—C11	119.5 (3)
O5—Cu2—O4	85.16 (11)	C17—C16—C14	118.8 (4)
O5—Cu2—N3	162.75 (13)	C17—C16—H16A	120.6
O4—Cu2—N3	94.60 (12)	C14—C16—H16A	120.6
O5—Cu2—N4	93.60 (12)	C16—C17—C18	120.9 (4)
O4—Cu2—N4	164.18 (13)	C16—C17—H17A	119.6
N3—Cu2—N4	81.93 (13)	C18—C17—H17A	119.6
O5—Cu2—C12	105.20 (9)	N2—C18—C17	121.8 (3)
O4—Cu2—C12	101.42 (9)	N2—C18—H18A	119.1
N3—Cu2—C12	91.78 (10)	C17—C18—H18A	119.1
N4—Cu2—C12	94.13 (10)	O4—C19—C20	125.1 (3)
C1—O1—Cu1	109.2 (2)	O4—C19—C24	117.4 (3)
C6—O2—Cu1	109.6 (2)	C20—C19—C24	117.5 (3)
C3—O3—C4	119.9 (3)	C21—C20—C19	119.2 (4)
C19—O4—Cu2	109.7 (2)	C21—C20—H20A	120.4
C24—O5—Cu2	109.4 (2)	C19—C20—H20A	120.4
C21—O6—C22	120.5 (3)	O6—C21—C20	123.0 (4)
C7—N1—C11	118.5 (3)	O6—C21—H21A	118.5
C7—N1—Cu1	128.6 (3)	C20—C21—H21A	118.5
C11—N1—Cu1	112.4 (2)	O6—C22—C24	120.5 (3)
C18—N2—C15	117.8 (3)	O6—C22—C23	113.1 (3)



C18—N2—Cu1	129.3 (3)	C24—C22—C23	126.3 (4)
C15—N2—Cu1	112.4 (2)	C22—C23—H23A	109.5
C25—N3—C29	118.6 (3)	C22—C23—H23B	109.5
C25—N3—Cu2	128.5 (3)	H23A—C23—H23B	109.5
C29—N3—Cu2	112.7 (2)	C22—C23—H23C	109.5
C36—N4—C33	118.2 (3)	H23A—C23—H23C	109.5
C36—N4—Cu2	129.4 (3)	H23B—C23—H23C	109.5
C33—N4—Cu2	112.4 (2)	O5—C24—C22	122.9 (3)
O1—C1—C2	124.9 (3)	O5—C24—C19	117.9 (3)
O1—C1—C6	117.7 (3)	C22—C24—C19	119.2 (3)
C2—C1—C6	117.4 (3)	N3—C25—C26	122.1 (4)
C3—C2—C1	119.4 (4)	N3—C25—H25A	119.0
C3—C2—H2A	120.3	C26—C25—H25A	119.0
C1—C2—H2A	120.3	C27—C26—C25	119.6 (4)
O3—C3—C2	123.1 (3)	C27—C26—H26A	120.2
O3—C3—H3A	118.4	C25—C26—H26A	120.2
C2—C3—H3A	118.4	C26—C27—C28	119.2 (3)
C6—C4—O3	120.9 (3)	C26—C27—H27A	120.4
C6—C4—C5	126.1 (4)	C28—C27—H27A	120.4
O3—C4—C5	113.0 (3)	C29—C28—C27	117.2 (4)
C4—C5—H5A	109.5	C29—C28—C30	118.5 (4)
C4—C5—H5B	109.5	C27—C28—C30	124.4 (3)
H5A—C5—H5B	109.5	N3—C29—C28	123.3 (4)
C4—C5—H5C	109.5	N3—C29—C33	116.2 (3)
H5A—C5—H5C	109.5	C28—C29—C33	120.5 (3)
H5B—C5—H5C	109.5	C31—C30—C28	121.2 (3)
O2—C6—C4	123.1 (3)	C31—C30—H30A	119.4
O2—C6—C1	117.7 (3)	C28—C30—H30A	119.4
C4—C6—C1	119.2 (3)	C30—C31—C32	121.3 (4)
N1—C7—C8	122.3 (4)	C30—C31—H31A	119.4
N1—C7—H7A	118.8	C32—C31—H31A	119.4
C8—C7—H7A	118.8	C33—C32—C34	116.9 (3)
C9—C8—C7	119.6 (4)	C33—C32—C31	118.6 (4)
C9—C8—H8A	120.2	C34—C32—C31	124.5 (4)
C7—C8—H8A	120.2	N4—C33—C32	124.1 (3)
C8—C9—C10	119.3 (4)	N4—C33—C29	116.0 (3)
C8—C9—H9A	120.4	C32—C33—C29	119.8 (3)
C10—C9—H9A	120.4	C35—C34—C32	118.4 (4)
C11—C10—C9	117.1 (3)	C35—C34—H34A	120.8
C11—C10—C12	118.4 (3)	C32—C34—H34A	120.8
C9—C10—C12	124.5 (4)	C34—C35—C36	120.8 (4)
N1—C11—C10	123.1 (3)	C34—C35—H35A	119.6
N1—C11—C15	116.2 (3)	C36—C35—H35A	119.6
C10—C11—C15	120.6 (3)	N4—C36—C35	121.5 (3)
C13—C12—C10	120.9 (3)	N4—C36—H36A	119.2
C13—C12—H12A	119.6	C35—C36—H36A	119.2
O2—Cu1—O1—C1	-3.4 (3)	C12—C10—C11—C15	0.0 (6)
N1—Cu1—O1—C1	-168.6 (3)	C11—C10—C12—C13	-0.1 (6)
N2—Cu1—O1—C1	-91.0 (6)	C9—C10—C12—C13	-179.7 (4)

## supplementary materials

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Cl1—Cu1—O1—C1	100.8 (2)	C10—C12—C13—C14	-1.7 (6)
O1—Cu1—O2—C6	4.3 (3)	C12—C13—C14—C15	3.4 (6)
N1—Cu1—O2—C6	99.0 (5)	C12—C13—C14—C16	-177.6 (4)
N2—Cu1—O2—C6	170.8 (3)	C18—N2—C15—C14	0.8 (6)
Cl1—Cu1—O2—C6	-96.1 (2)	Cu1—N2—C15—C14	-171.9 (3)
O5—Cu2—O4—C19	4.9 (3)	C18—N2—C15—C11	-177.9 (4)
N3—Cu2—O4—C19	167.6 (3)	Cu1—N2—C15—C11	9.4 (4)
N4—Cu2—O4—C19	91.1 (5)	C16—C14—C15—N2	-1.2 (6)
Cl2—Cu2—O4—C19	-99.6 (2)	C13—C14—C15—N2	177.9 (4)
O4—Cu2—O5—C24	-5.7 (2)	C16—C14—C15—C11	177.5 (3)
N3—Cu2—O5—C24	-95.7 (4)	C13—C14—C15—C11	-3.4 (6)
N4—Cu2—O5—C24	-169.9 (3)	N1—C11—C15—N2	-1.3 (5)
Cl2—Cu2—O5—C24	94.8 (2)	C10—C11—C15—N2	-179.5 (3)
O2—Cu1—N1—C7	-105.4 (5)	N1—C11—C15—C14	179.9 (3)
O1—Cu1—N1—C7	-12.0 (4)	C10—C11—C15—C14	1.7 (6)
N2—Cu1—N1—C7	-178.7 (4)	C15—C14—C16—C17	0.3 (6)
Cl1—Cu1—N1—C7	89.2 (3)	C13—C14—C16—C17	-178.8 (4)
O2—Cu1—N1—C11	83.0 (5)	C14—C16—C17—C18	0.8 (6)
O1—Cu1—N1—C11	176.4 (3)	C15—N2—C18—C17	0.4 (6)
N2—Cu1—N1—C11	9.7 (3)	Cu1—N2—C18—C17	171.7 (3)
Cl1—Cu1—N1—C11	-82.5 (3)	C16—C17—C18—N2	-1.3 (6)
O2—Cu1—N2—C18	12.1 (4)	Cu2—O4—C19—C20	176.2 (3)
O1—Cu1—N2—C18	98.7 (6)	Cu2—O4—C19—C24	-3.1 (4)
N1—Cu1—N2—C18	177.9 (4)	O4—C19—C20—C21	-177.4 (4)
Cl1—Cu1—N2—C18	-92.9 (3)	C24—C19—C20—C21	1.8 (6)
O2—Cu1—N2—C15	-176.2 (3)	C22—O6—C21—C20	0.3 (6)
O1—Cu1—N2—C15	-89.6 (6)	C19—C20—C21—O6	-1.4 (6)
N1—Cu1—N2—C15	-10.4 (3)	C21—O6—C22—C24	0.2 (6)
Cl1—Cu1—N2—C15	78.8 (3)	C21—O6—C22—C23	179.6 (3)
O5—Cu2—N3—C25	100.3 (5)	Cu2—O5—C24—C22	-174.0 (3)
O4—Cu2—N3—C25	11.8 (4)	Cu2—O5—C24—C19	5.8 (4)
N4—Cu2—N3—C25	176.3 (4)	O6—C22—C24—O5	-179.9 (3)
Cl2—Cu2—N3—C25	-89.8 (4)	C23—C22—C24—O5	0.9 (6)
O5—Cu2—N3—C29	-82.9 (5)	O6—C22—C24—C19	0.3 (6)
O4—Cu2—N3—C29	-171.4 (3)	C23—C22—C24—C19	-178.9 (4)
N4—Cu2—N3—C29	-7.0 (3)	O4—C19—C24—O5	-1.8 (5)
Cl2—Cu2—N3—C29	87.0 (3)	C20—C19—C24—O5	178.9 (3)
O5—Cu2—N4—C36	-11.7 (4)	O4—C19—C24—C22	178.0 (4)
O4—Cu2—N4—C36	-96.7 (5)	C20—C19—C24—C22	-1.3 (6)
N3—Cu2—N4—C36	-174.9 (4)	C29—N3—C25—C26	2.1 (6)
Cl2—Cu2—N4—C36	93.9 (3)	Cu2—N3—C25—C26	178.7 (3)
O5—Cu2—N4—C33	171.0 (3)	N3—C25—C26—C27	-2.5 (6)
O4—Cu2—N4—C33	86.0 (5)	C25—C26—C27—C28	0.8 (6)
N3—Cu2—N4—C33	7.8 (3)	C26—C27—C28—C29	1.0 (6)
Cl2—Cu2—N4—C33	-83.5 (3)	C26—C27—C28—C30	-178.5 (4)
Cu1—O1—C1—C2	-177.5 (3)	C25—N3—C29—C28	-0.1 (6)
Cu1—O1—C1—C6	1.9 (4)	Cu2—N3—C29—C28	-177.2 (3)
O1—C1—C2—C3	177.2 (4)	C25—N3—C29—C33	-177.8 (3)
C6—C1—C2—C3	-2.2 (6)	Cu2—N3—C29—C33	5.1 (4)

C4—O3—C3—C2	0.2 (6)	C27—C28—C29—N3	-1.5 (6)
C1—C2—C3—O3	1.8 (6)	C30—C28—C29—N3	178.1 (4)
C3—O3—C4—C6	-1.8 (6)	C27—C28—C29—C33	176.1 (4)
C3—O3—C4—C5	178.3 (3)	C30—C28—C29—C33	-4.3 (6)
Cu1—O2—C6—C4	176.1 (3)	C29—C28—C30—C31	2.7 (6)
Cu1—O2—C6—C1	-4.5 (4)	C27—C28—C30—C31	-177.7 (4)
O3—C4—C6—O2	-179.3 (3)	C28—C30—C31—C32	2.2 (6)
C5—C4—C6—O2	0.6 (7)	C30—C31—C32—C33	-5.5 (6)
O3—C4—C6—C1	1.3 (6)	C30—C31—C32—C34	173.4 (4)
C5—C4—C6—C1	-178.8 (4)	C36—N4—C33—C32	-2.0 (6)
O1—C1—C6—O2	1.8 (5)	Cu2—N4—C33—C32	175.7 (3)
C2—C1—C6—O2	-178.8 (3)	C36—N4—C33—C29	175.1 (3)
O1—C1—C6—C4	-178.8 (4)	Cu2—N4—C33—C29	-7.3 (4)
C2—C1—C6—C4	0.6 (6)	C34—C32—C33—N4	1.8 (6)
C11—N1—C7—C8	0.9 (6)	C31—C32—C33—N4	-179.2 (4)
Cu1—N1—C7—C8	-170.3 (3)	C34—C32—C33—C29	-175.1 (3)
N1—C7—C8—C9	0.3 (6)	C31—C32—C33—C29	3.9 (6)
C7—C8—C9—C10	-0.6 (6)	N3—C29—C33—N4	1.5 (5)
C8—C9—C10—C11	-0.3 (6)	C28—C29—C33—N4	-176.2 (4)
C8—C9—C10—C12	179.3 (4)	N3—C29—C33—C32	178.7 (4)
C7—N1—C11—C10	-1.9 (6)	C28—C29—C33—C32	1.0 (6)
Cu1—N1—C11—C10	170.6 (3)	C33—C32—C34—C35	-0.1 (6)
C7—N1—C11—C15	179.9 (3)	C31—C32—C34—C35	-179.0 (4)
Cu1—N1—C11—C15	-7.5 (4)	C32—C34—C35—C36	-1.4 (6)
C9—C10—C11—N1	1.6 (6)	C33—N4—C36—C35	0.4 (6)
C12—C10—C11—N1	-178.0 (4)	Cu2—N4—C36—C35	-176.8 (3)
C9—C10—C11—C15	179.7 (4)	C34—C35—C36—N4	1.3 (6)

Fig. 1

